

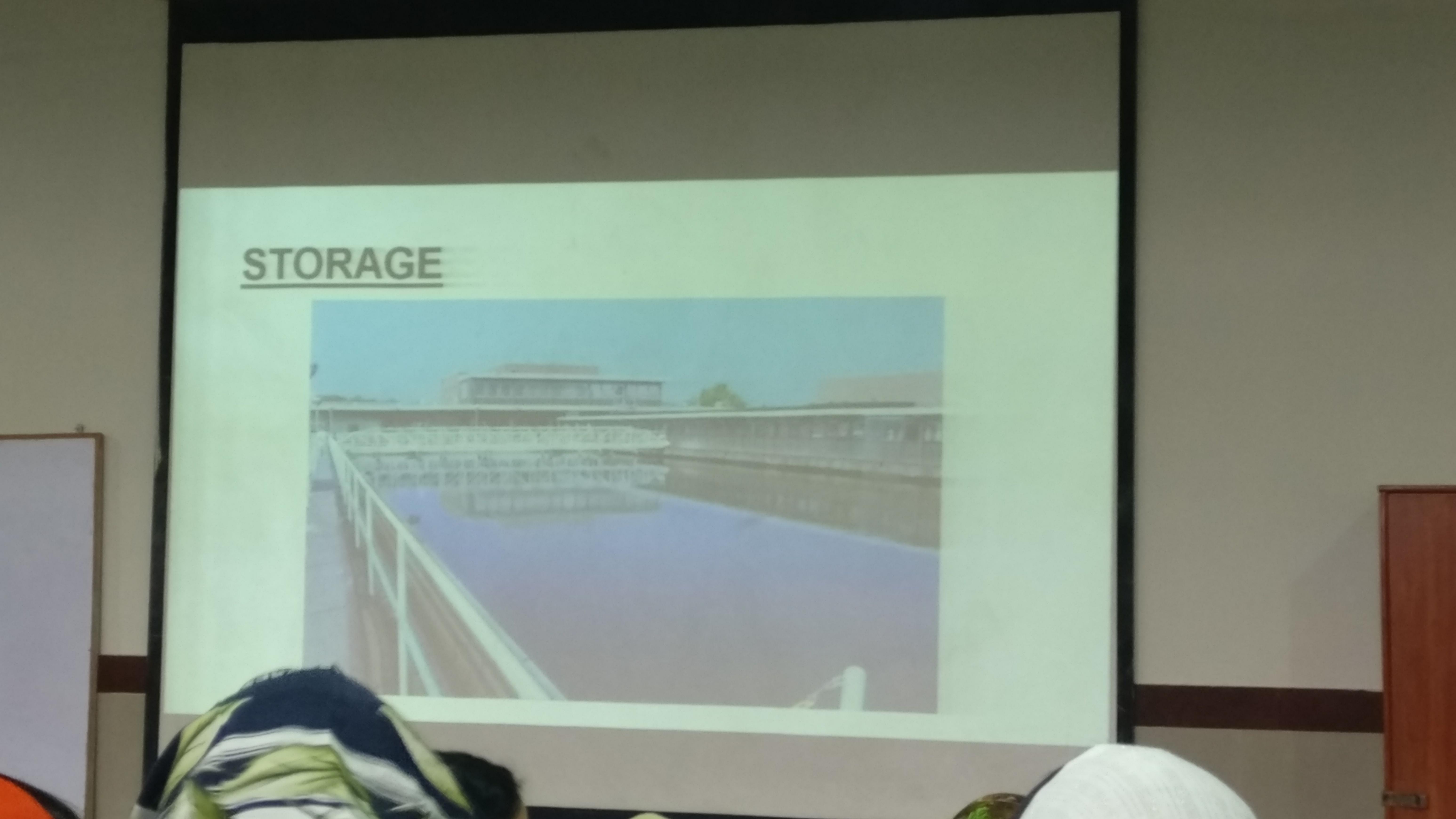
PURIFICATION ON LARGE SCALE

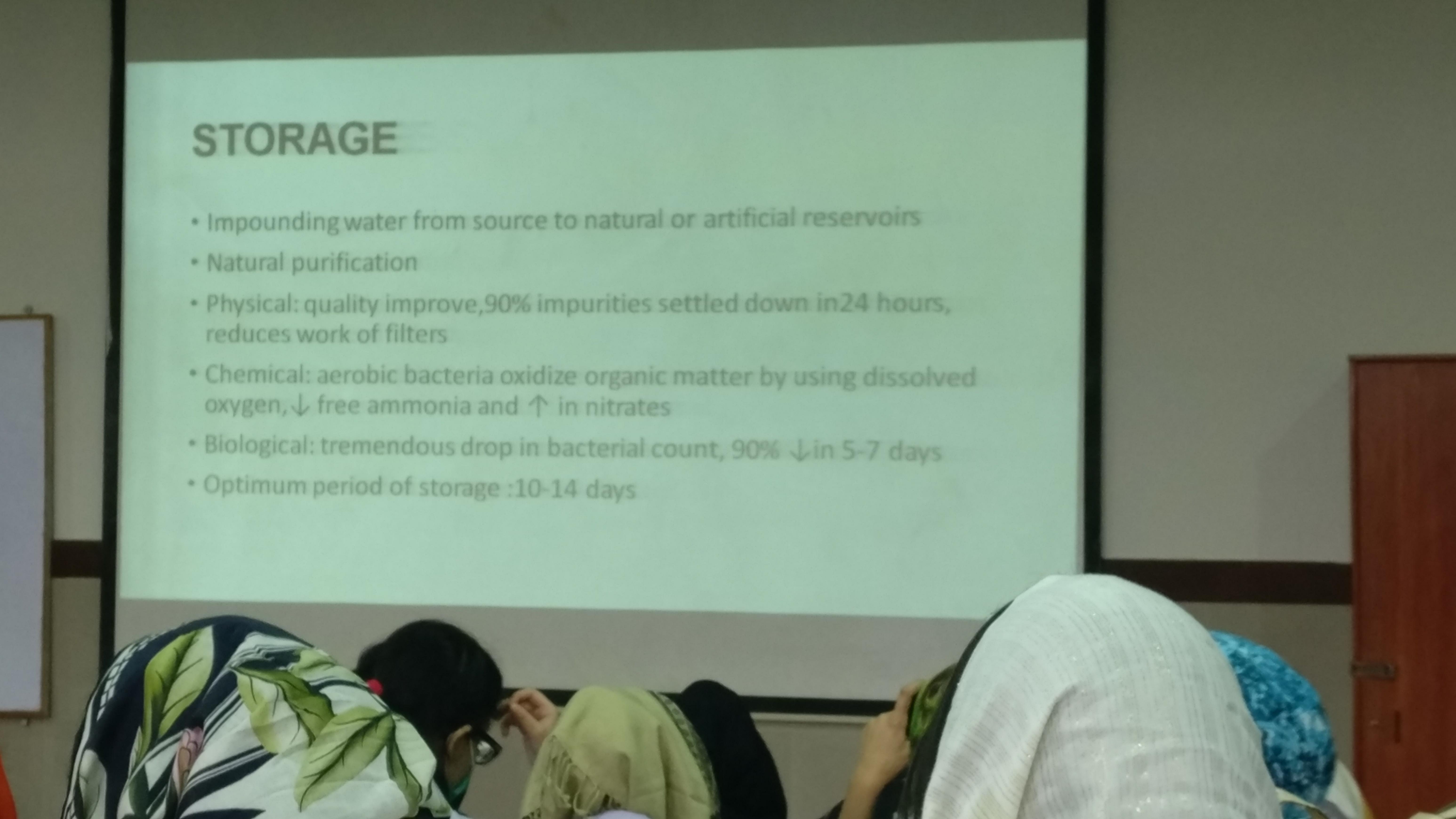
When sources of water are rivers, streams, dams, lakes and sea etc. then water is purified by the following methods

1/Storage

2/ Filtration

- * slow sand or biological filters
- * rapid sand or mechanical filters
- 3/ Disinfection
- · Chlorination: Water is disinfected by the addition of chlorine after filtration.





FILTRATION

- * Second stage of purification
- 98-99% bacteria removed by filtration
- *Twotypes

SLOW SAND OR BIOLOGICAL FILTERS:

Elements: a/supernatant water

b/a bed of graded sand

c/an under drainage system

d/a system of filter control valves



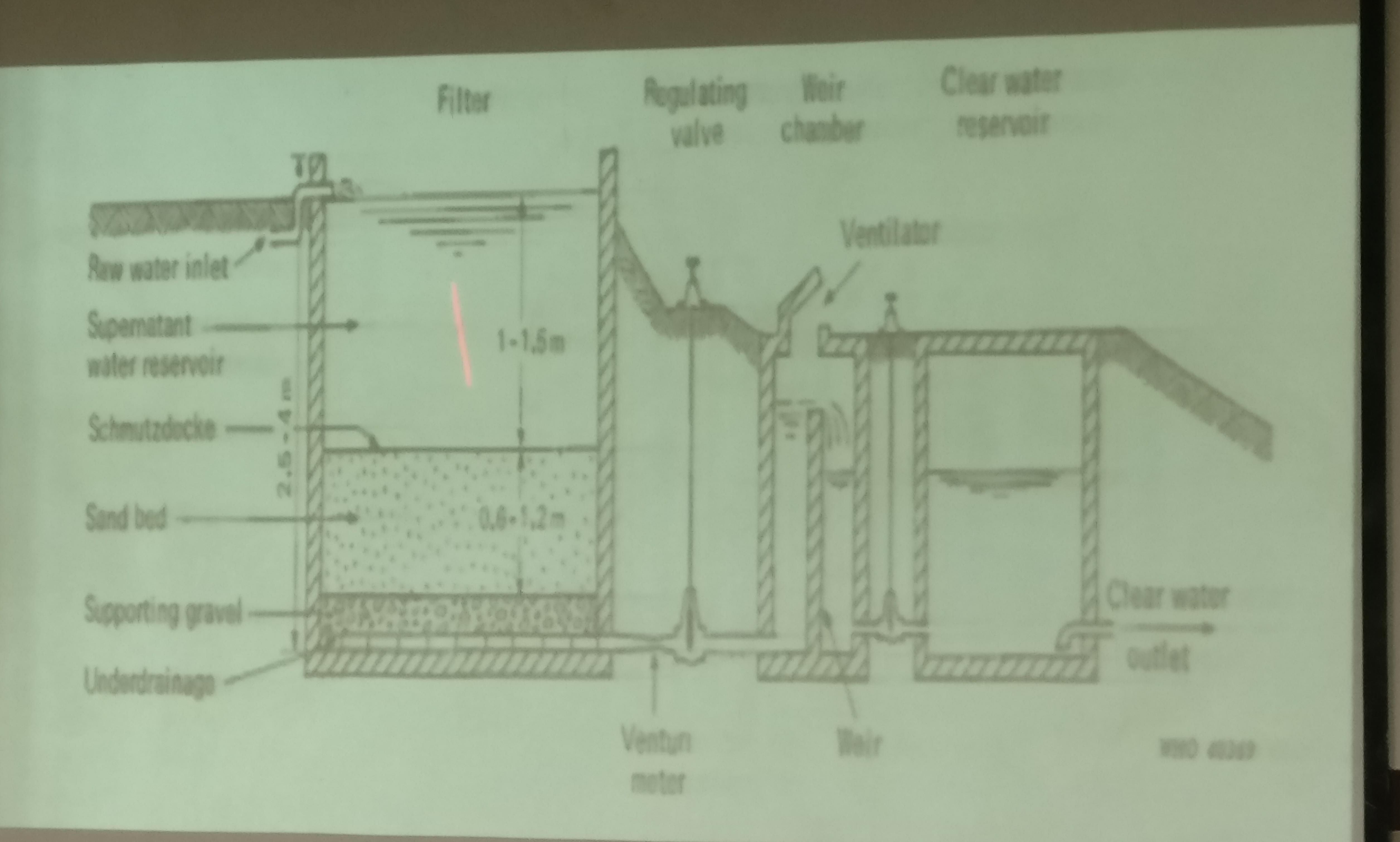
a/Supernatant water: depth

2 important purposes i.e. constant head → to overcome the resistance of filter bed, Secondly provide waiting period of 3-12hours for raw water

b/Sand bed: most important part

sand grains supported by graded gravel, water percolates through sand bed very slowly 2-more hours(purification through mechanical straining, sedimentation, oxidation and bacterial action





D/FILTER CONTROL

Venturi meter- measures bed resistance or loss of head

Filter cleaning- supernatant water drained off & by scraping of sand bed, thickness reduce to 0.5-0.8m, new bed is constructed

ADVANTAGES: simple, cheap, high quality filter water, reduce bacterial count 99%



RAPID SAND FILTER OR MECHANICAL FILTER

2 types:
gravity(Paterson)
&
pressure(candy)

Steps:
Coagulation,
Rapid mixing,
Flocculation,
Sedimentation
Filtration

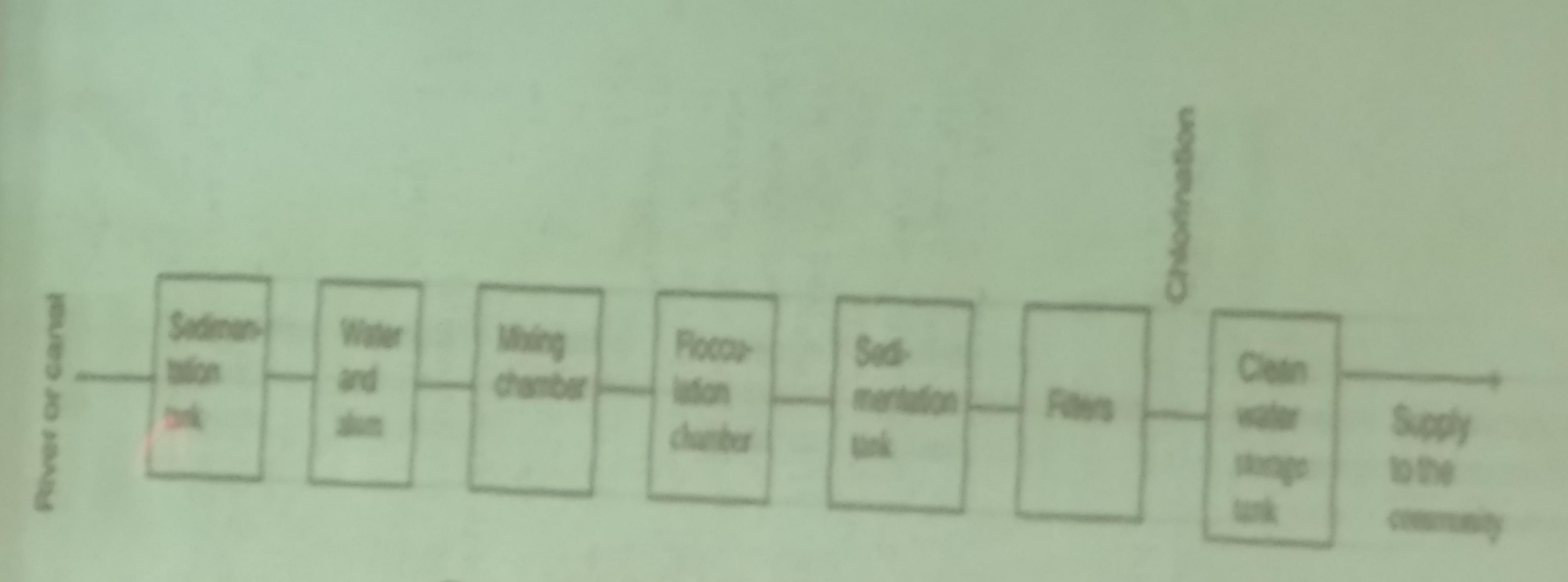


Fig 5.4: Flow diagram of rapid sand threaton plant

Comparison of Rapid & Slow Sand Filters.

	Slow Sand Filter	Rapid Sand Filter
	Occupies large space	Occupies very little area
		5-15 m ³ m ² h
Effective size of sand	0.15-0.35 mm	0.6 - 2.0 mm
Preliminary treatment	Plain sedimentation	Chemical coagulation
Washing	By Scraping the sand bed	By back-washing
Operations	Less skilled	Highlyskilled
Removal of turbidity	Good	Good
Removal of colour	fair	Good
Removal of backs	99.9-99.99 per cent	

DISINFECTION

Qualities

Chlorination

Action of chlorine

Principles of chlorination

Break point chlorination

Super chlorination

Orthotoliane test(07

Other agents and ozonation

Disinfection of wells

- · By bleaching powder(chlorinated lime CaOCL)
- Steps: a/ find the volume of water in a well:

Measure the depth of water column.....(h)m

Measure the diameter of well.....(d)m

Take the average of several readings of the above measurements.

* Substitute h and d in: Volume(liters)=3.14×d₂×h × 1000

4

• One cubic meter= 10000 liters of water

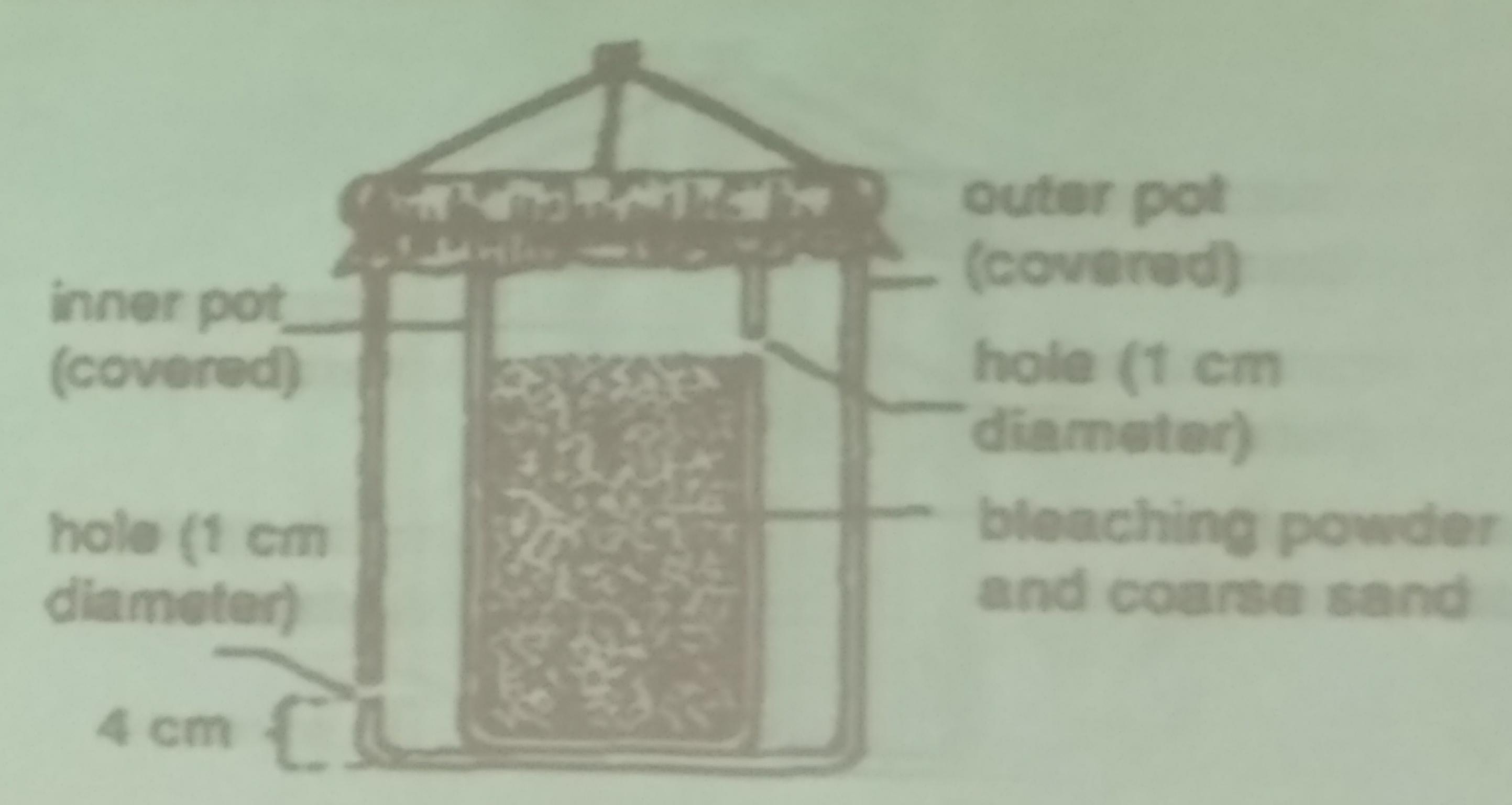
b/ find the amount of bleaching powder required for disinfection (2.5gms for 1000 liters of water)

Double Pot Method

Disinfection of Wells during an Emergency – The Double Pot Method
 E.g. an outbreak of cholera

. During an emergency, it is desirable to ensure a constant dosage of

chlorine to well water





ACTION OF CHLORINE

• When chlorine is added to water, it forms Hydrochloric Acid (HCL) & Hypochlorous Acid (HOCI). The disinfecting action of chlorine is mainly due to hypochlorous acid and to a small extent due to hypochlorite ions.



PRINCIPLES OF CHLORINATION

- 1. Water to be chlorinated must be free from turbidity.
- 2. Chlorine demand of water should be estimated.
- 3. Contact period for one hour is essential to kill the bacteria and viruses.
- 4. Minimum recommended concentration for free chlorine is 0.5 mg per liter for one hour.

Actual dose of Chlorine:

It is the sum of chlorine demand plus free residual chlorine

Chlorine Demand:

It is the difference between the amount of chlorine added to the water and the amount of residual chlorine remaining at the end of a specific period of contact (1 hour) at a given temperature and pH of water.

Residual Chlorine:

Amount of untreated chlorine, remaining in the water after some time as an effective disinfecting agent i.e. 0.2 ppm

Break point chlorination:

The point at which the chlorine demand of water is met and if further chlorine is added free chlorine begin to appear in water.

Super Chlorination:

It is addition of large doses of chlorine to the water and removal of excess of chlorine after disinfection.

Break point chlorination

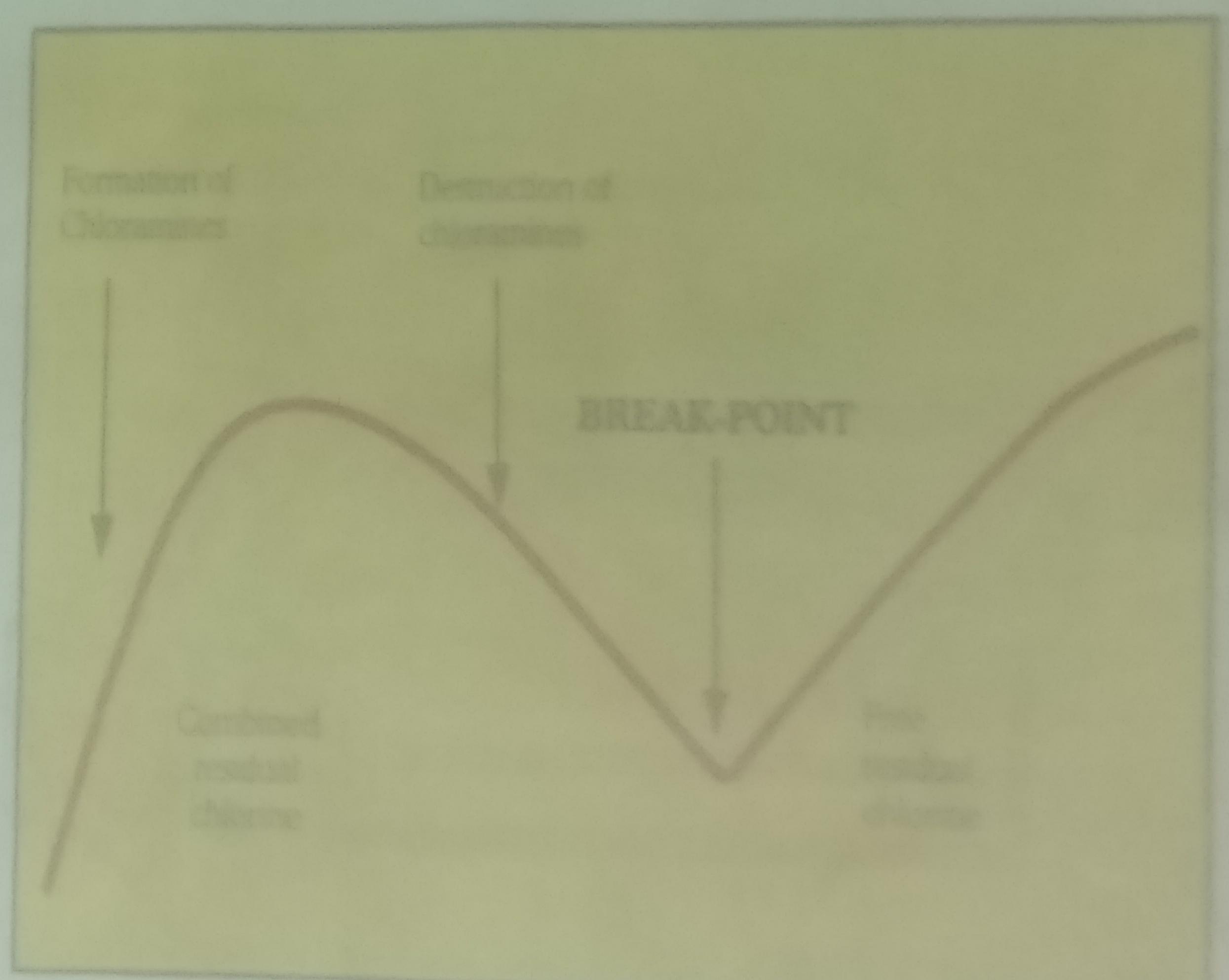
Break point chlorination is a systematic and scientific process and involves several steps. There are essentially four phases of chlorination:

Phase I: Formation of chloramines

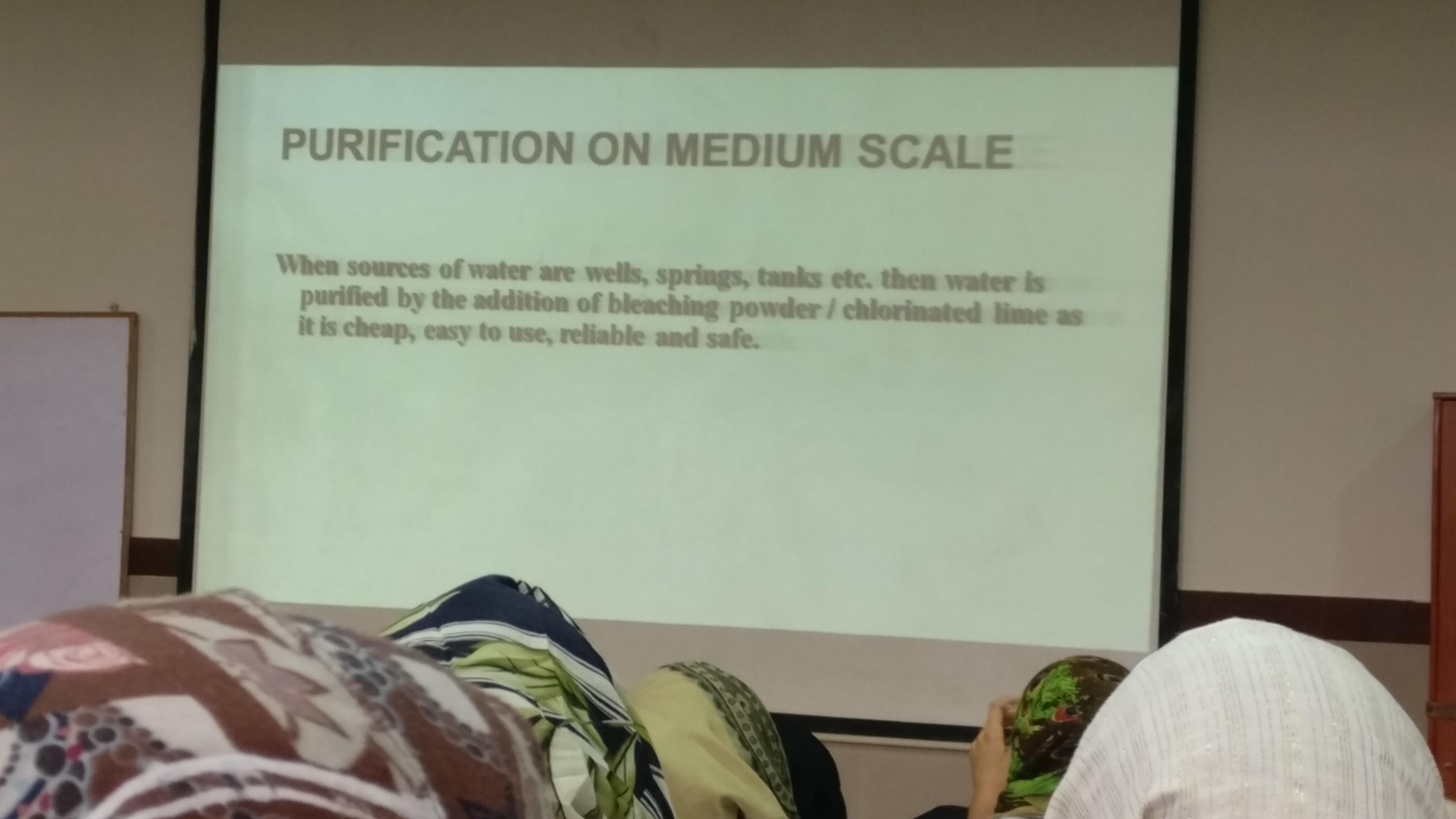
Phase II: Destruction of chloramines

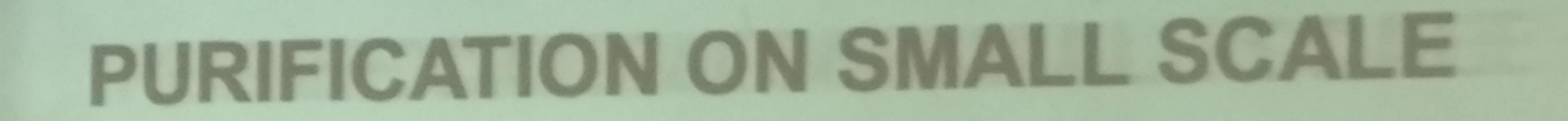
Phase III: Appearance of break-point

Phase IV: Accumulation of free residual chlorine









(DOMESTIC METHODS)

Boiling

Distillation

Addition of Chemicals

- * Bleaching Powder
- · Chiorine
- * lodine solution
- * KMnO,
- · Alum

4. Filtration